

SYNTHESIS OF 4-AMINO-4-(S)-
DIHYDROSPECTINOMYCIN

Sir:

Over the past decade, extensive studies have been done in the chemical modification of spectinomycin (**1**), in an effort to improve its biological activity.¹ Recent studies in this area have also resulted in two total syntheses^{2,3} of this unique structure and the establishment of the stereochemical identity⁴ of spectinoic acid, a rearrangement product of spectinomycin⁵.

A recent patent⁶ has disclosed the preparation of 4-amino-4-(R)-dihydropectinomycin from the parent antibiotic. This product has shown unexpectedly improved biological activity against a variety of microorganisms. We wish to report our concurrent studies along the same lines and the synthesis of 4-amino-4-(S)-dihydropectinomycin (**2**) from 4-(S)-dihydropectinomycin (Scheme 1).

Treatment of the known N,N'-dibenzoyloxycarbonyl dihydropectinomycin⁷ (**3**) with methanesulfonylchloride in pyridine (1.3 equiv. 0°C, 18 hours) gave the corresponding methanesulfonate derivative **4** as a chromatographically homogeneous amorphous solid (79%); $[\alpha]_D \pm 2.6^\circ$ (CHCl₃); m/z 584.236 (M⁺-CH₃SO₃H). Treatment of **4** with excess silver carbonate in

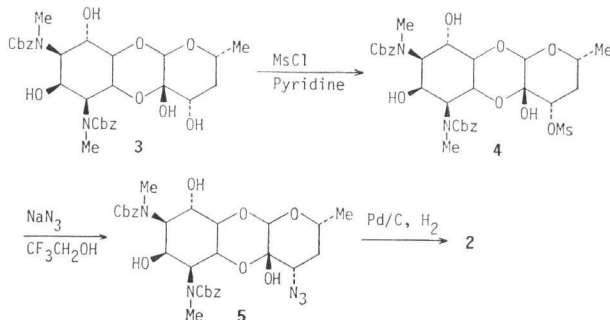
1:1 aq. acetone (reflux, 18 hours) gave the known^{7,8} N,N'-dibenzoyloxycarbonyl-4-(R)-dihydropectinomycin (47%, identical with authentic material), and N,N'-dibenzoyloxycarbonyl-actinamine as a byproduct.

Treatment of **4** with sodium azide (15 equiv.) in 1.5% aq. trifluoroethanol (reflux, 4 hours), gave after chromatographic separation, a product which was characterized as N,N'-dibenzoyloxycarbonyl-4-azido-4-(S)-dihydropectinomycin (**5**) (amorphous solid, 68% based on recovered starting material, 30% isolated) $[\alpha]_D + 20.1^\circ$ (CHCl₃) (Scheme 2). Catalytic hydrogenation (10% Pd/C, H₂, EtOH, dil. HCl) gave 4-amino-4-(S)-dihydropectinomycin (**2**) as a colorless solid (92%), mp 198~200°C (dec.); $[\alpha]_D + 11.4^\circ$ (H₂O); m/z 334 (M⁺+H); 316 (M⁺-H₂O) *etc.* ¹H n.m.r. data at 400 MHz (D₂O): p.p.m. 1.35 (d, 3H, C-2 Me, $J=6.2$ Hz), 1.77 (dd, 1H, H_{3ax}, $J_{3ax,3eq}=12.8$ Hz; $J_{3ax,4ax} \simeq J_{3ax,2ax}=12.7$ Hz), 2.06 (m, 1H, H_{3eq}, $J_{3eq,4ax}=4.9$ Hz; $J_{3eq,2ax}=1.7$ Hz), 2.88 (s, 6H, N-Me), 3.32 (dd, 1H, H₆, $J_{6,5a}=10.1$ Hz; $J_{6,7}=2.9$ Hz), 3.61 (dd, 1H, H₈, $J_{8,9}=11.0$ Hz, $J_{8,7}=2.8$ Hz), 3.71 (dd, 1H, H₄), 3.95 (m, 1H, H-2), 4.03 (dd, 1H, H-5_a, $J_{5a,9a}=10.1$ Hz), 4.11 (dd, 1H, H-9_a, $J_{9a,9}=10$ Hz), 4.37 (dd, 1H, H-9), 4.79 (dd, 1H, H-7), 4.96 (s, 1H, H-10_a). The above assignments were confirmed by decoupling experiments and it was conclusively estab-

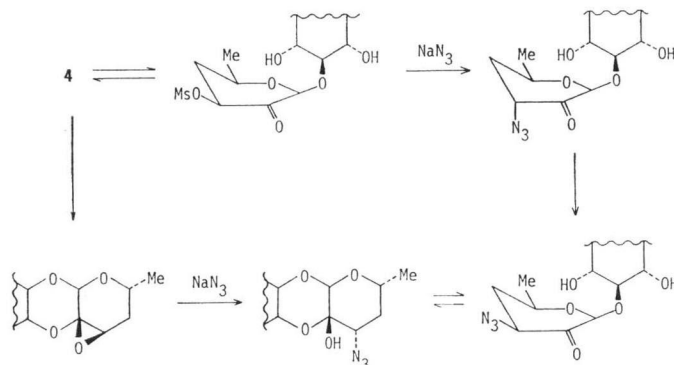
Scheme 1.



Scheme 2.



Scheme 3.



lished that the 4-amino group had an equatorial disposition, contrary to expectations based on an S_N2 type displacement of the mesylate and the solvolysis described above. Although the physical constants of the 4-(*S*)- and 4-(*R*)⁶⁾ amino derivatives are quite similar, their n.m.r. spectra and chromatographic properties were different.*

Compound **5** is therefore formed from **4** by a mechanism that promotes *retention* of configuration at C-4. We suggest two possibilities to account for this process as illustrated in Scheme 3. The mesylate **4** could in fact react as the α -keto mesylate which would be much more reactive⁹⁾ as compared to the bicyclic form **4**, and it would also be sterically favorable to an S_N2 type displacement reaction.** The azido derivative thus produced can now undergo epimerization and subsequent diastereoselective intramolecular acetal formation as expected of such a derivative^{9,10)} to give the observed product **5**. Alternatively the trans disposition of the tertiary hydroxyl group and the mesylate function could initially lead to epoxide formation, which could undergo regio- and stereoselective opening with azide ion. Other possibilities as proposed for the reaction of α -keto triflates¹¹⁾ could also be considered but may not be operative in this case.

To the best of our knowledge, the C-4 position in the dihydrospectinomycins has so far remained

* We thank Dr. R. MAIER, K. THOMAE for sending us a 250 MHz ¹H n.m.r. spectrum of 4-amino-4-(*R*)-dihydrospectinomycin.

** Models show that backside approach at C-4 is not favored in the conformationally rigid tricyclic structure **4**.

unexploited.⁶⁾ The mechanistic outcome of the displacement of the novel 4-(*S*)-mesylate derivative of dihydrospectinomycin may have relevance in future work aimed at introducing other functionality at that position. It should be noted however that 4-(*S*)-derivatives containing leaving groups such as in **4** are prone to stereoelectronically-controlled rearrangements, involving the antiperiplanar C-4a-C-10a bond¹²⁾ similar to other GROB-type fragmentations.¹³⁾ Unlike the 4-amino-4-(*R*)-dihydrospectinomycin (spectinomyclamine)⁶⁾ the synthetic 4-(*S*)-analog was found to be devoid of antibacterial activity.

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* There are two major byproducts in the azide displacement reaction which could arise from such a rearrangement. These are under study.

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